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Evaluation of Aloe arborescens gel as new coating to maintain the organoleptic and functional properties of strawberry (Fragaria × ananassa cv. Cadonga) fruits

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Evaluation of Aloe arborescens gel as new coating to maintain the organoleptic and functional properties of strawberries (*Fragaria x ananassa* cv. Cadonga) fruits

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338x190mm (96 x 96 DPI)

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3 1 **Running head: Strawberry preservation with *Aloe* gel coating**
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7 3 **Evaluation of *Aloe arborescens* gel as new coating to maintain the organoleptic and functional**
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9 **properties of strawberries (*Fragaria x ananassa* cv. Cadonga) fruits**
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32 33 15 **Summary**

34
35 16 **Strawberries (*Fragaria x ananassa* cv. Cadonga) are highly perishable fruits with a storage life, which**
36
37 may be less than a week. In this study, *Aloe arborescens* gel was used as postharvest treatment in
38 17
39 order to maintain strawberry quality. **Strawberries coated with edible *A. arborescens* gel were**
40 18
41 **packaged in a polypropylene box and stored.** Fruit titratable acidity, pH, soluble solid content, ascorbic
42 19
43 acid, total phenols, total flavonoids, total anthocyanins and antioxidant activity evaluated by two
44 20
45 different tests (DPPH and ABTS) were measured during 14 days of storage. Significant differences
46 21
47 **were** found ($p < 0.05$) for the samples treated with *A. arborescens* compared to the control. During
48 22
49 conservation, use of *Aloe* gel maintained lower values for total soluble solids, a higher concentration
50 23
51 of total phenols and ascorbic acid, and a better antioxidant activity when compared to the control.
52 24
53 The anthocyanin content remained largely unchanged throughout, in all compared samples.
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27 **Keywords:** Strawberry; Edible coating; *Aloe arborescens* gel; Postharvest quality.

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29 Introduction

30 The deterioration of fruits **after harvesting** are mainly due to the interactions between the fruits
31 and its surroundings, which can result in loss of moisture and of some compounds ([Andrade Pizarro
32 et al., 2016](#)). The strawberry (*Fragaria × ananassa*) is a shrub belonging to the Rosaceae family
33 ([Kim et al., 2011](#)) The fruit is soft and can be consumed fresh **or** processed in puree, juice or jam.
34 These fruits are **a low-calorie** (36 calories × 100 g) source of bioactive compounds including
35 polyphenols and vitamin C ([Soares, 20014](#)). **Strawberries** have a very short **shelf-life** due both to the
36 high respiration rate and to the absence of peel or rind and, therefore, strawberries rapidly lose water
37 and weight ([Pelayo et al., 2003](#); [Parvez & Wani, 2018](#)). For this reason, fresh, refrigerated fruits
38 should be consumed within 2-3 days after manual picking. Strawberries are usually packed on-site in
39 plastic punnets with lids.

40 To lengthen the **shelf-life** of fresh fruit, a correct choice of packaging must be taken into account
41 ([Sicari et al., 2017](#); [Rizzo et al., 2018](#); [Giuffrè et al., 2019](#)). Today, there are a variety of packaging
42 **methods**, some of which use innovative materials and techniques. A common aspect of modern
43 packaging is to offer “convenience” and ease of use, combined with a minimal environmental impact
44 ([Sicari et al., 2017](#); [Rizzo et al., 2018](#); [Giuffrè et al., 2019](#)). In recent years, research has been directed
45 towards edible coatings, which can fulfil many of the above-mentioned requirements. Edible coatings
46 are **materials, which** mimic the natural external layer of fruit and vegetables ([Del Valle et al., 2005](#);
47 [Chien et al., 2007](#); [Tzoumaki et al., 2009](#)). These coatings have many advantages: they are
48 biodegradable materials that can be eaten together with the produce, and reduce the environmental
49 impact by wholly, or in part, replacing synthetic packaging with particular reference to plastics. These
50 coatings provide a barrier to moisture **and** oxygen, and consequently they reduce weight loss during
51 the fruit’s **shelf-life** ([Akhtara et al., 2015](#)). The application of edible coating can be **carried out** by
52 brushing, spraying or dipping ([Mchugh & Senesi, 2000](#)). **Furthermore, to edible coatings additives**
53 **may be added, especially antimicrobial, to help extend the shelf-life of the produce, as has been**

1
2
3 54 widely reported in the literature (Krasaekoopt & Mabumrung, 2008; Rojas-Graü *et al.*, 2008).
4
5 55 Furthermore, several studies propose the addition of texture enhancers to edible coatings to minimize
6
7 56 softening during storage of fresh-picked fruits (Rojas-Graü *et al.*, 2008). Recently, *Aloe vera* gel has
8
9
10 57 gained much attention for use as an environmentally friendly and safe post-harvest treatment.
11
12 58 It has been proposed as a coating for fresh fruit since it does not change the food's taste and
13
14 59 appearance (Misir *et al.* 2014) and can be used as a completely normal and useful alternative to
15
16 60 artificial preservatives. Valverde, *et al.* (2005) and Martínez-Romero *et al.* (2006), used it as a coating
17
18 61 for cherries and table grapes, and it was shown not only to be useful in reducing microbial
19
20 62 proliferation, but also to have a positive effect on reducing moisture loss, on maintaining firmness,
21
22 63 and in checking respiration. A similar effect was observed also for peach, nectarine, guava, plum and
23
24 64 kiwi fruits (Guillén *et al.*, 2013; Hazrati *et al.*, 2017). More recently apart from *A. vera* gel, the gels
25
26 65 obtained from other *Aloe* spp. such as *A. arborescens* and *A. ferox* have been investigated for this
27
28 66 purpose. Taking into account the increased consumer interest in ready-to-eat fruit with high
29
30 67 nutritional value, we wished to evaluate the protective effect of the application of *Aloe arborescens*
31
32 68 gel as edible coating on strawberry fruits. The effectiveness of *A. arborescens* gel, used as an edible
33
34 69 coating, was evaluated and compared to a widely used, commercially available, polypropylene-based
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36 70 anti-fog film (PP). Untreated and unpackaged strawberries were used as a control.
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44 72 **Materials and methods**

45 73 **Chemicals and Materials**

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47 74 Folin-Ciocalteu reagent, 2,2-diphenyl-1-picrylhydrazyl radical (DPPH⁺) and 2,2'-azinobis-(3-
48
49 75 ethylbenzothiazoline-6-sulfonic acid) (ABTS⁺), were supplied by Carlo Erba (Milan, Italy). Solvents
50
51 76 and reagents not expressly specified had a high degree of purity and were supplied by Carlo Erba
52
53 77 (Milan, Italy), gallic acid, 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox) and
54
55 78 aluminium chloride (AlCl₃), were supplied by Aldrich (Milan, Italy). Anti-fog film (PP). was supplied
56
57 79 by Cartonpack, Rutigliano, BA, Italy.

80

81 **Preparation of *Aloe arborescens* gel**

82 *Aloe arborescens* leaves used for the production of the gel were harvested from plants, approximately
83 6 years old, growing wild in the province of Reggio Calabria (Italy). The leaves chosen were neither
84 the old leaves from the bottom of the plant, nor the new leaves from the top, but rather were chosen
85 from the middle part of the plant. The fresh leaves were picked in the early hours of the morning and
86 immediately taken to the Food Technology laboratory, where they were washed with 2% v/v
87 chlorinated water (sodium hypochlorite). After drying, to extract the gel, the parenchyma was ground
88 in a commercial blender. The obtained matrix was filtered to remove the fibrous fraction, under
89 vacuum filtration with Buchner funnel using laboratory filter paper. The gel was pasteurized in a
90 thermostatic bath at 70 °C (core) for 45 minutes (Arowora *et al.*, 2013), and allowed to cool
91 immediately to room temperature before further use. The treatment was monitored using a Data
92 Logger (Escort Junior, Astori tecnica, Italy). After pasteurization and cooling, the gel was stored in
93 brown glass bottles to prevent oxidization.

94

95 **Fruit selection and treatments**

96 *Fragaria x ananassa* cv. Cadonga fruits were manually picked in the early hours of the morning in
97 May 2017 from a farm in the province of Reggio Calabria (Italy). They were harvested at a
98 commercially mature stage, sorted to eliminate damaged, shrivelled, and unripe fruit, and selected for
99 uniform size and color. Following this screening, the berries were washed with chlorinated water
100 (sodium hypochlorite 2% v/v), rinsed with distilled water, carefully dried with paper towels, and
101 placed in plastic punnets (15 cm x 6 cm x 6 cm), each of them containing around 250 g of strawberries
102 (about eighteen fruits). Forty-two of these containers were filled, of which fourteen samples had no
103 treatment other than being washed in chlorinated water (C, control), fourteen sample containers were
104 sealed using a commercially-available polypropylene-based anti-fog film (PP), and fourteen samples
105 contained fruit treated with *Aloe arborescens* gel (EC). The treatment with EC was carried out by

immersing fruits for 1 min at room temperature in *Aloe arborescens* gel. Three replicates for each treatment were conducted. Based on our previous studies, longer dipping times showed no change in results (data not shown). After dipping, the fruits were placed on stainless steel trays and air-dried for 60 minutes at 15 °C. All forty-two samples, subdivided as described, were refrigerated at 3 ± 1 °C (90 % RH).

Monitoring of shelf-life parameters

Changes in the shelf-life parameters such as weight loss, whole-fruit firmness, were monitored at time 0, 2, 5, 7, 9, 12, and 14 days of storage. All the analyses were carried out in triplicate.

Weight loss measurement

To determine weight loss, strawberries were weighed by using Analytical Balance ML54T/00 (Mettler Toledo S.p.A. Milan, Italy) at the beginning of the experiment and after coating ($t=0$), and thereafter each sampling during the storage period. Weight loss was expressed as the percentage loss of the initial total weight. Ten fruits in three repetitions were used to evaluate the weight loss.

Fruit colour surface measurement

The strawberries' surface colour was measured at room temperature using Konica Minolta CM-700/600d spectrophotometer (Konica Minolta Sensing, Inc., Japan). CIE L^* a^* b^* parameters were recorded. The values a^* and b^* were used to calculate the Chroma value (C) using the following equation: $C = \sqrt{a^{*2} + b^{*2}}$

Whole-fruit firmness measurement

Fruit firmness was measured as total firmness by a compression test. TA.XT PLUS Texture Analyser (Stable Micro Systems, Godalming, UK) was used, which profiled a mechanical force displacement using a 50 kg loading cell and equipped with a 5 mm diameter flat probe. Fruit firmness values

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3 132 were an average of 6 strawberries. Each sample was subjected to a two-cycle compression with 5 s
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5 133 between cycles. The highest value of force required to compress the sample during the first
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7 134 compression cycle was recorded as fruit firmness. This test measured individual fruit firmness based
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10 135 on the resistance of the flesh to deformation by the probe. Firmness was measured at the equatorial
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12 136 part and on two faces of each fruit. Three repeated measurements were performed for each sample
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15 137 and the results were expressed in Newton (N).
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19 139 **Determination of pH, titratable acidity, total soluble solid and ascorbic acid content**

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21 140 Strawberries (c.a. 5 g) were ground in a commercial blender, 50 mL of H₂O was added before
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24 141 homogenization using Ultraturrax T-25 (Ika Labortechnik, Janche & Kunkel, Milan, Italy). The mix
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26 142 was centrifuged at 5000 rpm for 10 min and the supernatant was collected and used for all
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28 143 determinations.
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31 144 A Crison basic 20 pH meter (Crison Instruments S.A., Milan, Italy) was used to test the pH of
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33 145 strawberry samples. Titratable acidity (TA) results were expressed as citric acid percentage on dry-
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35 146 weight basis. Strawberry soluble solid content (TSS) was measured at 20 °C using a digital Atago
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37 147 Model PR-101 α refractometer (Atago Co. Ltd, Milan, Italy). Results were reported as Brix degrees
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40 148 (°Brix). All determinations above described were carried out in triplicate. The concentration of
41
42 149 ascorbic acid in strawberries was measured by the method reported by [Thimmaiah, \(1999\)](#). Results
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44
45 150 were expressed as mg of ascorbic acid/100 g of fresh weight (FW).
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49 152 **Strawberry extraction procedure**

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51 153 Fresh strawberries (10 g) were homogenized (1 min at 24000 rpm) with 30 mL of
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54 154 methanol/water/hydrochloric acid (80:19.9:0.1, % v/v) solution using Ultraturrax T-25 homogeniser
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56 155 (Ika Labortechnik, Janche & Kunkel, Milan, Italy), and further extracted at room temperature under
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58 156 continuous stirring for 1 h in the dark. The residue obtained by vacuum-filtration (Whatman n. 1 filter,
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60 157 Vetrotecnica Srl, Padova, Italy) was re-extracted three times (until colourless) under the same

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3 158 conditions to maximize the antioxidant recovery. The filtrates were combined, evaporated to dryness
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5 159 using a rotary evaporator, and dissolved in a methanol/water mixture for further analysis.
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10 161 **Evaluation of total phenols and flavonoids content**

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12 162 The total polyphenol content (TPC) of *Strawberry* fruit was determined by the Folin-Ciocalteu
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14 163 method as previously reported (Singleton *et al.*, 1999). The absorbance was measured at $\lambda= 760$ nm.
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16

17 164 The results were expressed as gallic acid equivalents (GAE) in mg/100 g FW.
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19 165 The total flavonoid content (TFC) in strawberry extract was measured by the aluminium chloride
20
21 166 colourimetric assay as previously reported (Zhishen *et al.*, 1999). The absorbance was measured at
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23 167 510 nm. The total flavonoid content was expressed as mg catechin equivalents (CE)/100 g of fresh
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25 168 weight (FW). All analyses were carried out in triplicate.
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31 170 **Determination of total anthocyanins**

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33 171 Total anthocyanin content (TAC) was measured using the described method (Kara & Erçelebi, 2013).
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35 172 Absorbance was measured at 510 nm and 700 nm in buffers at pH 1.0 and pH 4.5. Experiments were
36
37 173 conducted in triplicate.
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40 174 41 42 175 **Evaluation of antioxidant activity**

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44 176 The antioxidant activity (AA) was screened by using 2,2-diphenyl-1-picrylhydrazyl (DPPH)
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46 177 and 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) methods.
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49 178 The DPPH test was assessed following the method reported by Brand-Williams *et al.* (1995). Briefly,
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51 179 DPPH· methanolic solution was added to strawberry extract. The absorbance at $\lambda= 515$ nm was
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53 180 measured. All tests were carried out in triplicate and the results expressed as means \pm standard
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55 181 deviation (SD).
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58 182 The ABTS (2,2'-azinobis-3-ethylbenzothiazoline-6-sulfonate) radical test was carried out as
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60 183 described by Re *et al.* (1999). Briefly, a solution of ABTS radical was diluted (1:80) with ethanol to

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184 give an absorbance of 0.70 at $\lambda = 734$ nm. An aliquot of extract was added to ABTS solution. Trolox
185 was used as a standard antioxidant and fruit activity was expressed in mM of Trolox equivalents.

10 187 **Statistical analysis**

12 188 Results were expressed as mean \pm SD of three replicates. All data were analyzed using one-way
13
14 189 analysis of variance (ANOVA) with SPSS 17.0 (SPSS Inc., Chicago, IL, USA) statistical
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16
17 190 software. Significant differences were calculated according to Duncan's multiple range tests.
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19 191 Differences at $P < 0.05$ were considered to be statistically significant while at $P < 0.01$ were considered
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21 192 to be highly significant. PCA was performed on the auto-scaled data matrix, and the principal
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23
24 193 components were extracted so that the dimensionality of the original data matrix was reduced while
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26 194 retaining the maximum variability. SPSS software was used for statistical analysis of PCA.

31 196 **Results and discussion**

33 197 **Weight loss**

35 198 Since strawberries have no peel or waxy substance on the skin, these fruits are susceptible to rapid
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37 199 moisture evaporation, resulting in softening, shrinking, surface wounding, and darkening of skin
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40 200 colour. The use of edible coatings such as *A. arborescens* gel makes a layer on strawberry skins that
41
42 201 reduces water loss, defends against mechanical damage and microbial attack, and also seals small
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44 202 wounds. The loss of strawberry weight during the shelf-life period represents not only a quality factor,
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46
47 203 but also a problem of economic impact for producing companies. As expected, the control sample
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49 204 shows the greatest weight loss (13.52%), whereas the samples treated with *A. arborescens* gel (EC)
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51 205 and polypropylene-based film (PP) show a loss of 1.16% and 0.19% respectively. The analysis of
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53
54 206 variance shows that the percentage of weight loss for strawberries coated with *Aloe arborescens* gel
55
56 207 was significant ($p < 0.01$) compared to the control. These results are in agreement with those of
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58 208 Mahmoud & Savello, (1992) and Avena-Bustillos *et al.* (1997) who concluded that coatings and/or
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60 209 films significantly conserved water content. Even if the samples in the polypropylene-based film (PP)

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3 210 show less weight loss (data not shown), it can be hypothesized that the aloe gel coating nonetheless
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5 211 was effective in reducing weight loss during refrigerated storage. This reduction in weight loss on the
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7 212 part of the strawberries coated with aloe gel is due to the gel's acting as a barrier to the movement of
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10 213 moisture, thus reducing its loss during post-harvest storage. Our results are better than those obtained
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12 214 by Singh *et al.* (2011), who used *A. vera* gel as edible coating on strawberries stored for 16 days and
13
14 215 obtained a weight loss of 9.99 % compared to the untreated fruit (13.79%). The superiority of *A.*
16
17 216 *arborescens* gel compared to *A. vera* gel is also demonstrated. In fact, Nasrin *et al* (2017) reported a
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19 217 weight loss of 10.67% in control and 3.68% in treated fruits after nine days of storage.
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21
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23 24 219 **Fruit firmness**

25
26 220 The fruit firmness, for all treatments, followed a declining trend commensurate with length of storage.
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28 221 However, PP and EC significantly maintained firmness compared with control (PP: 2.91 and EC:
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30 222 2.72 vs C: 1.83 after 9 days of storage). Interestingly, the firmness value is 3.3-time higher in EC
31
32 223 sample in comparison to the control at the end of the observation period. The analysis of variance
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34 224 shows that strawberry firmness for strawberries coated with *Aloe arborescens* gel was significant
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36 225 ($p < 0.01$) compared to the control. Recently, Nasrin *et al.* (2017) found that strawberries treated with
37
38 226 *A. vera* gel lost only 18.43% of firmness in comparison to uncoated fruits after 15 days of storage.
39
40 227 We can state that edible coating showed a good result for firmness probably because this coating
41
42 228 slowed down metabolism. These results were in agreement with those of Maftoonazad &
43
44 229 Ramaswamy, (2005) and Koh & Melton, (2002), who stated that retention of firmness could be
45
46 230 explained by retarded degradation of insoluble protopectins to the more soluble pectic acid and pectin.
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48
49 231

50 51 232 **Physical chemical parameters**

52
53 233 Table 1 shows the values of the analyses carried out on the strawberries in three different storage
54
55 234 conditions (C, PP, EC). Titratable acidity (TA), total soluble solids (TSS), pH and colour were
56
57 235 monitored over the 14 days of storage at 3 ± 1 °C. The untreated samples (C) had a pH of 3.45 when

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2
3 236 freshly picked (t=0). The addition of *A. arborescens* gel (EC) or storage under film (PP) did not cause
4
5 237 significant differences to pH when compared to the control (C). Generally, no significant modification
6
7
8 238 in pH was observed during the period of observation in all investigated samples. **There were**
9
10 239 **significant differences for AT, TSS and colour among different treatments during the storage of 14**
11
12 240 **days (Table 1). The effects of coating treatments on the TA and TSS parameters during storage are**
13
14 241 **shown in Table 1.** The value for titratable acidity was 0.49 g/L at t=0. This value increased for all
15
16
17 242 three storage types (C, PP, EC) although the greatest increase was in the control sample at t=14, which
18
19 243 had a value of 0.98 g/L, whereas the samples stored using *Aloe arborescens* gel or stored in PP film
20
21 244 had very similar values (0.74 and 0.71 g/L respectively). **The TA levels in the control and coated**
22
23 245 **samples gradually decreased during the storage period, and the difference was significant in the**
24
25 246 **control sample only on day 9. However, the decreasing trends of TA in coated samples were not**
26
27 247 **significant during the storage period.** These results are in disagreement with those reported by [Vahdat](#)
28
29 248 [et al. \(2010\)](#), that found a high TA value in *A. vera* gel treated strawberries in comparison with
30
31 249 untreated fruits. **The TSS of the control, coated gel and PP film significantly increased with storage**
32
33 250 **time, while the coated and PP film samples showed a slight increase compared to the control sample**
34
35 251 **(Tab. 1).**
36
37
38 252 **In fact,** at t=0 strawberries showed a total soluble solids (TSS) value of 7.20 °Brix. Fruits treated with
39
40 253 *A. arborescens* gel or film (PP) maintained a value of 8.20 °Brix, whereas, for the control, the value
41
42 254 of TSS reached 11.20 °Brix at t=14, **similar to the values reported by [Benítez et al., 2013](#).**
43
44
45 255 This **increase in TSS** could be due to the hydrolysis of structure polysaccharides, and the release of
46
47 256 simple compounds (sugars). This hypothesis could be corroborated by the loss of consistency, due to
48
49 257 the hydrolysis of the same polymers. However, **these** data are in disagreement with those reported by
50
51 258 [Nasrin et al. \(2017\)](#) and [Sing et al. \(2011\)](#), **who** found a moderate increase of TSS content in *A. vera*
52
53 259 gel treated fruits. The use of *Aloe* gel as a coating does not seem to influence *L** parameter. For
54
55 260 Chroma, the three sets of values show a similar trend for the first 7 days of storage while a drastic
56
57 261 reduction of C value was observed after 12 days of storage in all samples, although the sample treated

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2
3 262 with EC presents a lower reduction of Chroma compared to PP and C (Table 1). This reduction in
4
5 263 chromatic coordinates correspond to a dark red color on the skin of the fruit as commonly happens
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7
8 264 when strawberries become fully ripe. The application of *A. arborescens* gel coating maintains the
9
10 265 fruit's colour intensity, whereas the control or the fruit in polypropylene-based film show a more
11
12 266 marked change of colour. Certainly, the absence of any protection for the control fruit accelerated the
13
14
15 267 fruit's post-harvest metabolic aging. The data obtained are similar to those in the literature regarding
16
17 268 the use of different edible coatings for strawberries. In particular, our data are in agreement with those
18
19 269 for strawberries treated with *A. vera* gel (Ribeiro *et al.*, 2017). A reduction in C and L* values was
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21
22 270 observed for combined chitosan-oleic acid edible coated strawberries refrigerated at 5 °C for 10 days
23
24 271 (Vargas *et al.*, 2006).

272 273 **Bioactive compound content**

274 Epidemiological studies showed that strawberry consumption has been associated with several health
275
276 benefits probably due the antioxidant activity of its bioactive constituents (Forbes-Hernández *et al.*,
277
278 2016; Giampieri *et al.*, 2017). For this purpose we decided to monitor the content of bioactive
279
280 constituents including ascorbic acid content (AC), total polyphenol content (TPC), total flavonoid
281
282 content (TFC), and total anthocyanin content (TAC) during strawberry shelf-life. Results are reported
283
284 in Table 2. Sapei & Hwa, (2014) found that freshly-picked strawberries have a high and extremely
285
286 variable ascorbic acid content, usually between 5 and 50 mg/100 g of FW. At t=0 we found the ascorbic
287
288 acid content was 56.37 mg/100 g, reaching at t=14 values of 56.39, 65.79, and 74.59 mg/100 g for C,
289
290 PP and EC respectively. The analysis of variance shows that *Aloe arborescens* gel coating and PP
291
292 film on ascorbic acid of strawberry (Tab. 2) were significant ($p < 0.01$) compared to the uncoated fruit.
293
294 . As reported in the literature (Weichmann *et al.* 1985; Salunkhe *et al.*, 1991), the oxidation of vitamin
295
296 C is carried out by ascorbic oxidase and phenoloxidase, which have a low affinity for oxygen.
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298 Ascorbic acid content decreased during storage for uncoated strawberries (C). *Aloe arborescens* gel
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300 coating was effective in reducing the ascorbic acid loss (Tab. 2). The application of *Aloe* gel coating

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3 288 preserved AA content, and our results are in line with those reported by Singh *et al.* (2011) and Nasrin
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5 289 *et al.* (2017) however both these studies used *A. vera* gel as fruit coatings.
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8 290 The total phenolic compounds were affected by the treatments and the storage time. In Table 2 it can
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10 291 be seen that until t=2 total polyphenol values do not show significant variations ($p<0.01$) for all three
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12 292 treatments (C, PP, EC), whereas from t=5 the total polyphenol content in C differs significantly from
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14 293 PP and EC. At t=0 the strawberries showed a TPC value of 97.83 mg/100 g FW. During storage, the
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17 294 TPC in samples C and PP increased to t=5 then decreased to t=14 giving a final value of 68.14 and
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19 295 85.62 mg/100 g FW for samples C and PP respectively. For those strawberries treated with *A.*
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21 296 *arborescens* gel (EC) the TPC increased to t=5 (112.52 mg/100g) and t= 7 (115.24 mg/100g), before
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24 297 slowly decreasing. This may be due to the continued biosynthesis of the compounds after harvesting.
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26 298 However, unlike samples C and PP, the final TPC for EC was very close to the t=0 value.
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29 299 As opposed to what was found for total polyphenols and vitamin C, the total anthocyanin content
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31 300 showed a slight increase in the control sample, from 18.54 mg/100 g FW at t=0 to 18.22 mg/100 g
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33 301 FW at t=14 days, peaking at 24.23 and 24.12 mg/100 g FW at t=5 and t=7, respectively. For the
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35 302 sample treated with aloe gel, however, total anthocyanins increased to 21.41 mg/100 g at t=14 days.
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38 303 The anthocyanin of treated fruit increased during cold storage, similar to those reported previously
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40 304 (Hassanpour, 2015), that may be due to the continued biosynthesis of the compounds after harvesting.
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42 305 This trend can be explained by the barrier effect of the gel, which modifies the fruit's metabolism by
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45 306 reducing respiration (Martínez-Romero *et al.*, 2006). Flavonoids followed a similar pattern with total
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47 307 phenolics, having higher values after 7 days shelf life (Table 2). Total flavonoid content at t=0 was
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49 308 236.22 mg/100 g FW. Over the first five days, a sharp increase was observed, especially in
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51 309 strawberries coated with *A. arborescens* gel. At t=14, samples C and EC showed a significant increase
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54 310 with highest content in EC (652.96 mg/100 g FW) whereas strawberries in polypropylene-based film
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56 311 (PP) showed a smaller increase, although still greater than at t=0. Probably because this coating
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58 312 reduces metabolism.
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314 **Antioxidant activity**

315 Table 2 reports the results of the antioxidant capacity of strawberry extracts by DPPH and
316 ABTS assays. Obtained data, expressed in mM Trolox equivalents, show that EC samples, during the
317 experiment, have higher antioxidant activity compared to the others (C and PP). In particular, at t=14
318 days the antioxidant capacity of EC strawberry extract was 0.92 mM Trolox for DPPH test and 8.46
319 mM Trolox for ABTS assay. From our results, it is possible to see how the samples treated with *A.*
320 *arborescens* gel show a much greater content of polyphenols and flavonoids compared to samples C
321 and PP. Similar results have been found by Gol *et al.*, (2013). Therefore, we can say that the coating
322 reduces damage to the strawberries, stimulating a scavenging activity towards the radicals DPPH and
323 ABTS.

325 **Principal Component Analysis**

326 Results were analyzed by a multivariate PCA method to reach a smaller number of artificial variables
327 accounting for most of the variance in the observed variables (D'Agostino *et al.*, 2014).

328 The obtained results are shown in Fig. 1. In all considered samples, the first five PC accounted for
329 100% of total variance. Two PC's (PC1 and PC2), explaining the cumulative 80% of the data
330 variance, were chosen based on the Eigen values (> 1). The Scree plot shows the variance of each
331 component in the dataset, used to determine how many components should be retained in order to
332 explain a high percentage of the variation in the data. As can be seen in Fig. 1 most of the variance
333 in strawberries are explained by PC1 and PC2. PC1 explains about 45% and PC2 about 35% of the
334 total variance. Fig. 1 showed that the values of the analyses carried out at sampling t=12 and t=14 are
335 very close to each other and are characterized by a high value in component 2. The (t=0) showing
336 negative values on PC1 and PC2, while (t=2, t=5, t=7 and t=9), show positive values on PC1 and
337 PC2. In the analysis of the loading variable, a correspondence clearly appeared between PC1 and
338 TPC (C, EC and PP), ABTS (PP), Chroma (C, PP, EC), AA (C), TAC (C) and AC (PP).

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3 339 In particular, the analysis of the loading variables clearly showed the positive correlation of PC2 with
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5 340 AC (C and EC), TFC (C, PP and EC), AA (EC and PP), ABTS (PP), Brix (C and PP). Negative
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7 341 correlations were observed on PC1 and PC2 with DPPH (EC and PP), ABTS (C), and pH (EC and
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10 342 PP).

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14 344 **Conclusions**

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17 345 The application of bio-based edible coatings to prolong the **shelf-life** of fruits, which are in line with
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19 346 sustainable agriculture practices, is, at present, a hot research topic. *Aloe* gel coating has shown itself
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21 347 to be a promising method of conservation since it does not affect taste and appearance of the fruits.
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24 348 In this paper, we propose, for the first time the application of *Aloe arborescens* gel to fruits. Results
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26 349 showed that *A. arborescens* gel coatings may be an important way to conserve strawberries during
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28 350 refrigerated storage, and may be used instead of the plastic **films, which** are commercially available
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31 351 today. The species *A. arborescens* has been shown to be better than *A. vera* gel for this application.
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33 352 Therefore, this type of edible coating could become a useful treatment to improve quality and
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35 353 lengthen **shelf-life** of these widely-consumed fruits. However, further studies are necessary for the
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38 354 commercial application of *A. arborescens* gel in order to test different concentrations of gel find the
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40 355 optimum gel concentration for post-harvest treatments.

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44 357 **Abbreviations**

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47 358 **AA**, Ascorbic acid content; **ABTS**, 2,2'-Azinobis-(3-ethylbenzothiazoline-6-sulfonic acid; **C**,
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49 359 Control; **DPPH**, 1,1-Diphenyl-2-picryl-hydrazil; **EC**, Edible Coating with *Aloe arborescens* gel; **N**,
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51 360 Newton; **PP**, Polypropylene film; **TA**, Titratable acidity; **TAC**, Total Anthocyanins Content; **TFC**,
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53 361 Total Flavonoids Content; **TSS**, Total soluble solid content; **TPC**, Total Polyphenols Content.

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58 363 **Conflicts of interest**

59
60 364 Authors declare no conflict of interest.

365

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For Peer Review

Table 1 Change in strawberry fruit titratable acidity (TA), pH, total soluble solid content (TSS) during storage at 3°C, in control, PP film and *Aloe arborescens* coated.

Analysis		Shelf-life monitoring period (days)						
		0	2	5	7	9	12	14
TA	C	0.49±0.02 ^{aE}	0.74±0.03 ^{bC}	0.67±0.01 ^{cD}	0.98±0.06 ^{aA}	0.88±0.03 ^{aB}	0.91±0.04 ^{aB}	0.98±0.04 ^{aA}
	PP	0.49±0.02 ^{aE}	0.84±0.01 ^{aA}	0.81±0.02 ^{aB}	0.81±0.05 ^{bB}	0.67±0.02 ^{cD}	0.65±0.03 ^{cD}	0.71±0.06 ^{bC}
	EC	0.49±0.02 ^{aD}	0.81±0.03 ^{aB}	0.72±0.01 ^{bC}	0.95±0.04 ^{aA}	0.74±0.03 ^{bC}	0.74±0.02 ^{bC}	0.74±0.03 ^{bC}
pH	C	3.45±0.10 ^{aC}	3.29±0.02 ^{bD}	3.42±0.22 ^{aC}	3.13±0.17 ^{aE}	3.30±0.24 ^{aE}	3.75±0.12 ^{aA}	3.58±0.27 ^{aA}
	PP	3.45±0.11 ^{aB}	3.24±0.03 ^{bC}	3.33±0.16 ^{bC}	3.02±0.21 ^{aD}	3.35±0.28 ^{aC}	3.51±0.17 ^{bA}	3.56±0.33 ^{aA}
	EC	3.45±0.11 ^{aAB}	3.52±0.2 ^{aA}	3.33±0.18 ^{bBC}	3.04±0.26 ^{aD}	3.31±0.16 ^{aD}	3.46±0.16 ^{bAB}	3.61±0.41 ^{aA}
TSS	C	7.20±0.14 ^{aE}	7.22±0.55 ^{aE}	8.11±0.33 ^{aD}	9.10±0.22 ^{aC}	11.25±0.88 ^{aA}	10.10±0.88 ^{aB}	11.20±0.88 ^{aA}
	PP	7.20±0.2 ^{aD}	6.25±0.3 ^{aF}	7.12±0.52 ^{aE}	7.10±0.62 ^{aE}	9.20±0.41 ^{aA}	8.10±0.75 ^{aC}	8.20±0.96 ^{aB}
	EC	7.20±0.16 ^{aD}	7.30±0.41 ^{aC}	6.20±0.31 ^{aF}	6.20±0.55 ^{aF}	7.10±0.62 ^{aE}	8.10±0.67 ^{aB}	8.20±0.74 ^{aA}
Color	C	17,2±0.32 ^{aD}	16,62±0.56 ^{aE}	21,12±0.87 ^{bC}	23,64±1.65 ^{dA}	21,36±0.87 ^{fB}	8,83±0.58 ^{fF}	8.52±0.58 ^{bG}
	PP	17,2±0.31 ^{aE}	23,84±0.57 ^{aB}	20,51±0.69 ^{bD}	23,58±1.54 ^{eC}	27,79±1.06 ^{eA}	13,12±0.87 ^{eF}	12.88±0.69 ^{bG}
	EC	17,2±0.32 ^{aE}	22,12±1.03 ^{aD}	23,19±0.67 ^{abC}	23,34±1.06 ^{fB}	24,03±1.12 ^{eA}	13,73±0.96 ^{dF}	13.45±1.03 ^{bG}
L*	C	27,54±0.78 ^{aA}	25,42±0.98 ^{aA}	31,56±1.21 ^{aA}	27,98±2.05 ^{aA}	29,21±1.87 ^{bA}	30,24±2.44 ^{bA}	29.36±3.01 ^{aA}
	PP	27,54±0.74 ^{aAB}	26,52±0.89 ^{aB}	27,85±1.36 ^{aAB}	27,63±2.22 ^{bAB}	31,53±1.48 ^{aAB}	32,75±2.36 ^{aB}	31.02±2.87 ^{aAB}
	EC	27,54±0.76 ^{aD}	27,03±1.03 ^{aE}	26,15±2.04 ^{aBG}	26,59±2.31 ^{cF}	27,67±1.89 ^{dC}	29,74±3.65 ^{cA}	28.84±2.13 ^{aB}

Abbreviations: C: control; PP: polypropylene-based anti-fog film; EC: Edible coating; TA: titratable acidity; TSS: soluble solid content. Data are reported as mean ± standard deviation (n=3). Differences were evaluated by one-way analysis of variance (ANOVA) test completed with a multicomparison Tukey's test. ** $p < 0.05$ compared with the positive control. Means in the same row with different capital letters differ significantly ($p < 0.05$), means in the same column with different small letters differ significantly ($p < 0.05$).

Table 2 Change in strawberry fruit total phenols (TPC), total flavonoids (TFC), total anthocyanin (TAC) ascorbic acid (AA) and oxygen radical absorbance capacity during storage at 3°C, in control, PP film and *Aloe arborescens* coated.

Analysis	Shelf-life monitoring period (days)							
	0	2	5	7	9	12	14	
TPC								
GAE mg/100 g FW	C	97.83±3.21 ^{aB}	105.24±4.33 ^{aA}	108.03±5.36 ^{bA}	96.25±2.44 ^{cB}	88.24±3.02 ^{cC}	77.51±4.33 ^{cD}	68.14±3.55 ^{cE}
	PP	97.83±2.17 ^{aC}	102.44±4.67 ^{aB}	110.56±5.01 ^{abA}	105.36±5.39 ^{bB}	100.21±3.54 ^{bC}	89.68±4.21 ^{bD}	85.62±4.33 ^{bD}
	EC	97.83±2.66 ^{aD}	101.28±3.88 ^{aC}	112.52±4.87 ^{aA}	115.24±6.31 ^{aA}	104.36±2.88 ^{aB}	98.51±4.09 ^{aD}	94.36±4.30 ^{aD}
TFC								
CE mg/100g FW	C	236.22±3.56 ^{aF}	532.31±5.34 ^{cE}	664.32±2.98 ^{bC}	718.53±6.26 ^{bA}	690.40±4.22 ^{bB}	667.97±3.11 ^{aC}	636.34±4.22 ^{bD}
	PP	236.22±3.85 ^{aF}	573.26±5.62 ^{bB}	659.97±4.23 ^{bB}	662.50±6.07 ^{cB}	783.42±5.09 ^{aA}	532.32±2.69 ^{cD}	477.60±3.44 ^{cE}
	EC	236.22±4.09 ^{aE}	629.50±6.22 ^{aD}	794.83±3.67 ^{aA}	778.65±5.88 ^{aB}	645.30±5.22 ^{cC}	625.58±4.36 ^{bD}	652.96±5.22 ^{aC}
TAC								
CYE mg/100g FW	C	18.54±0.88 ^{aD}	22.14±0.77 ^{aB}	24.23±0.99 ^{aA}	24.12±1.36 ^{aA}	21.24±1.67 ^{aC}	20.36±1.66 ^{aC}	18.22±2.01 ^{cD}
	PP	18.54±1.23 ^{aC}	20.12±1.36 ^{bABC}	22.14±1.08 ^{bA}	21.41±1.57 ^{bAB}	22.01±2.09 ^{aAB}	21.98±2.07 ^{aA}	20.87±2.33 ^{bBC}
	EC	18.54±1.37 ^{aB}	18.67±1.64 ^{cB}	19.21±1.22 ^{cB}	18.54±1.64 ^{cB}	19.45±2.66 ^{bB}	19.99±2.03 ^{aAB}	21.41±2.38 ^{aA}
AA								
mg/g FW	C	56.37±2.56 ^{aB}	82.72±3.26 ^{aAB}	78.15±4.33 ^{bA}	63.64±2.64 ^{cAB}	74.63±3.45 ^{cAB}	69.34±3.54 ^{bAB}	56.39±4.21 ^{cA}
	PP	56.37±3.74 ^{aF}	73.53±4.21 ^{bC}	74.07±5.02 ^{cC}	81.82±2.09 ^{bB}	83.96±6.23 ^{bA}	63.87±3.19 ^{cE}	65.79±4.39 ^{bD}
	EC	56.37±3.69 ^{aF}	70.85±3.89 ^{cE}	88.89±4.77 ^{aA}	87.27±3.01 ^{aB}	85.30±4.21 ^{aC}	82.98±4.01 ^{aD}	84.59±3.85 ^{aC}
DPPH								
(mM Trolox)	C	1.37±0.02 ^{aA}	0.79±0.03 ^{cC}	0.88±0.01 ^{bB}	0.79±0.02 ^{cC}	0.74±0.01 ^{cD}	0.78±0.02 ^{bC}	0.67±0.01 ^{cE}
	PP	1.37±0.01 ^{aA}	0.84±0.02 ^{bC}	0.87±0.01 ^{bB}	0.83±0.02 ^{bC}	0.78±0.01 ^{bD}	0.72±0.01 ^{cE}	0.83±0.02 ^{bC}
	EC	1.37±0.02 ^{aA}	0.94±0.02 ^{aC}	0.89±0.02 ^{aE}	0.88±0.03 ^{aE}	0.93±0.02 ^{aC}	0.97±0.01 ^{aB}	0.92±0.01 ^{aC}
ABTS								
(mM Trolox)	C	8.10±0.65 ^{aA}	7.42±0.44 ^{cD}	7.62±0.77 ^{bC}	7.09±0.88 ^{cF}	7.21±0.88 ^{cE}	7.89±0.61 ^{bB}	7.51±0.44 ^{cD}
	PP	8.10±0.56 ^{aC}	8.23±0.36 ^{aB}	8.09±0.65 ^{aC}	8.29±0.74 ^{bC}	8.44±0.49 ^{bA}	7.87±0.59 ^{bD}	7.77±0.82 ^{bE}
	EC	8.10±0.38 ^{aE}	8.14±0.29 ^{bE}	7.59±0.56 ^{bF}	8.64±0.62 ^{aC}	9.23±0.61 ^{aA}	9.08±0.67 ^{aB}	8.46±0.93 ^{aD}

Abbreviations: C: control; PP: polypropylene-based anti-fog film; EC: Edible coating; TPC: total phenols content; TFC: total flavonoids content; TAC: total anthocyanins content; AA: vitamin C/ascorbic acid content. Data are reported as mean ± standard deviation ($n=3$). Differences were evaluated by one-way analysis of variance (ANOVA) test completed with a multicomparison Tukey's test. ** $p < 0.05$ compared with the positive control. Means in the same row with different capital letters differ significantly ($p < 0.05$), means in the same column with different small letters differ significantly ($p < 0.05$).

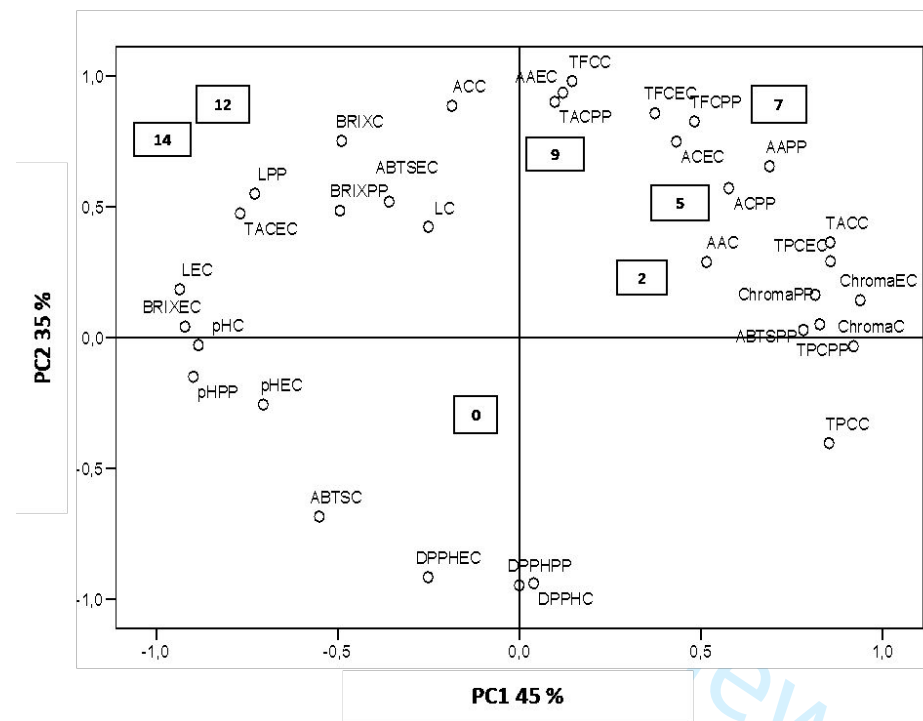


Fig. 1. Loadings and scores plots of principle components analysis (PC1 vs. PC2) on the main sources of variability between samples. PC1 first principle components and PC2 second principle components.

Referee: 1

Q1: The subject of the manuscript entitled “Evaluation of Aloe arborescens gel as new coating for maintain the organoleptic and functional properties of strawberry (*Fragaria x ananassa* cv. Cadonga) fruits” is innovative. The article is well written and describes the changes on fruit coated with Aloe arborescens gel stored at 3°C/14 days. The titratable acidity, pH, color, weight, soluble solid content, ascorbic acid, total phenols, total flavonoids, total anthocyanins, antioxidant activity by DPPH and ABTS test and texture were evaluated during storage. The materials and methods section is well described, but traditional spectrophotometric methods were used and only 2 methods “in vitro” were applied to determine the antioxidant activity. The results are discussed and the conclusions are supported by these results. To improve the article, I will suggest a more deeply discussion of the results, including more details about the characteristics of *A. arborescens* that could explain the behaviour of the coating, that is different of the *A. vera*. Also, it could be addressed a statistical analysis by principal components to set the relationship between the parameters analysed. Actually, the Pearson coefficient could not be the most appropriated statistical tool.

A1: Thank you so much for the comment. As suggested, principal component analysis (PCA) has been carried out, and there is a more detailed discussion of results. Following Refviewer’s suggestion we have decided to remove Pearson’s correlation coefficient calculation.

Q2: Other changes suggested are:

Line 58: remove the word “antioxidant”.

Line 208: add “data not shown”

Line 224: add “Table 1”

Line 319: replace “founded” by “found”

There are some results without significant letters in Table 1.

A2: The statistical analysis was completed in the table 1. Moreover, we have removed the word “antioxidant” from line 58; We have add: “data not shown” In line 208; We have inserted Table 1 in line 224; We have replace “founded” by “found” in line 319;

Referee: 2

Q1: Introduction. The authors develop an investigation based on the use of an Aloe gel to extend the useful life in strawberry, but do not consider more up-to-date information about it, specifically in relation to formulations and possible components in the formulation, plasticizers or others that improve the wetting angle and adhesion of the film, for example. In this case, it would be of interest to include updated information about the use of Aloe gels in formulations for coating in fruits. Then the minimally processed products, their loss of moisture and volatiles are pointed out, but these aspects are disconnected with the argumentation of the Introduction, even more so when the product to be treated is considered in its entirety. There are different format errors, statements not supported by bibliography Line 36 and line 38: without references ... a very large paragraph mentions only one Source, which is not correct.

A1: Thank you so much for the comment. Many works in the literature on the use of edible coatings suggest the use of additives such as antioxidants... to improve the product’s shelf-life.

1
2
3 In our experiment we decided to use 100% *aloe arborescens* to make an edible gel, since
4 almost all the literature uses *aloe vera* (also on strawberries). Thus, in order to evaluate fully
5 the effects of the *aloe arborescens* gel, we chose not to use additives. The part not regarding
6 the scientific aims has been eliminated. The references have been added.
7
8

9 Q2: Phytopathological and physiological breakdown problems are also outlined as the most
10 important problems in this type of fruit, the reader thinks that this was the focus of the article, but
11 it is not. Then there is a lack of bibliographical reference in Line 44 that supports his affirmation.

12 A2: The sentence has been improved and the following references have been added: To lengthen
13 the shelf-life of fresh fruit and vegetables a correct choice of packaging must all be taken into
14 account (Sicari *et al.*, 2017; Rizzo *et al.*, 2018; Giuffrè *et al.*, 2019).
15
16

17 Q3: In Line 47: requires bibliographic references that support your claim.

18 A3: Thank you so much for the comment. The references have been added
19
20

21 Q4: Line 53 contains only one reference for all parraf? Missing point in: "line 53 reference to plastics
22 (Andrade Pizarro *et al.*, 2016). These coatings provide a barrier to moisture ",
23

24 A4: Thank you for suggestion we have inserted the wrong reference the correct one is Akhtara,
25 J., Omreb, P. K., Ahmad Azad, Z.R.A. (2015). Edible coating for preservation of perishable foods:
26 a review. *J. Ready Eat Food.* 2, 81-88. This reference is now inserted.
27
28

29 Q5: The authors must define a format criteria for the definition of some terms related to the post-
30 harvest of the fruit, such as shelf-life or shelf life (Lines 54, 56, 276).
31

32 A5: The terms shelf life was replaced with "shelf-life" in all text.
33
34

35 Q6: Write again sentence in line 56: "Furthermore, edible coatings can contain additives,
36 especially antimicrobial, to help extend the shelf life of the production, as has been widely
37 reported in the literature (Krasaekoopt & Mabumrung, 2008), antioxidants (Rojas-Graü *et al.*,
38 2008);).
39

40 A6: the sentence was re-written.
41
42

43 Q7: To write again, the sentence in line 61 is not understood: "From Aloe two major liquid could be
44 obtained: the yellow latex and the colorless gel, which is obtained from parenchymal cells
45 (Chandegara & Varshney, 2013) which is used in several commercial drinks "
46

47 A7: We have removed this sentence and the correspondent reference.
48
49

50 Q8: In Material and Methods section: There is no important information to determine the scope
51 of the results obtained, no mention is made of the concentration used in the applied extract, no
52 prior experiments developed to the respect. Neither do they indicate the performance of the
53 extraction process or greater details of the pasteurization process. Greater background and details
54 of this stage of extraction and preparation of the gel are required. In Page 7, line 108: The authors
55 must explain which is the fundament of selection of treatment with EC treatment or if they count
56 with preliminary results that supports this decision.
57

58 A8: We thank for this comment. Necessary information was added to materials and methods
59 section.
60

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3 Q9: The authors not considered evaluation of effect of modification of the internal atmosphere
4 and gaseous exchange between fruit and external atmosphere like Valverde et al., (2015). Is
5 suggested to evaluate this parameter. It is necessary to detail the technical characteristics of gas
6 permeability of the material used (polypropylene PP boxes used). The authors mention using
7 polypropylene-based anti-fog film (PP-Cartonpack, Rutigliano, BA, Italy), should incorporate more
8 information regarding gas permeability.
9

10 A9: We thank for this comment. Unfortunately, in the experiment carried out, we did not
11 measure the internal atmosphere of the PP packages, since our primary objective was to evaluate
12 the effect of the edible coating on the fruit. We are aware that it would have been interesting to
13 add this measurement our work, and we will certainly bear this suggestion in mind in future
14 studies.
15
16

17
18 Q10: The authors said in line 110 that "The three different batches of containers (C, PP, EC) were
19 used" ... the sentence is not clear respect to the number of replicates for each treatment, 3
20 replicates for each treatment or they use one container for treatment?
21

22 A10: Three replicates for each treatment were used.
23

24 Q11: Details of the ripening stage of the fruits used in the experiment were not provided, nor do
25 they give information regarding the physiological state of the fruit used, degrees brix, acidity,
26 which greatly influences the possibilities of conservation of the product in a refrigerated
27 environment. 250 grams shows approximately 18 fruits in each plastic container. 42 plastic
28 containers, 14 samples without treatment, 14 in anti-Fog PP, 14 treated with Aloe.
29

30 A11: Many thanks for your careful observation. We have added the information in the text as
31 presented below: Strawberry fruit (*Fragaria x ananassa*) grown at Reggio Calabria (Italy) were
32 hand-harvested at a commercially mature stage, sorted to eliminate damaged, shriveled, and
33 unripe fruit, and selected for uniform size and color. After picking fruits were immediately
34 refrigerated at 4-5 °C and transferred to the laboratory in one hour and analyses were conducted.
35 The °Brix was determined in the orchard after picking.
36
37

38 Q12: It is not indicated the age of plants from where they collected Aloe leaves, the physiological
39 age influences the richness of the different components that the authors indicate that they
40 generate protective effects in the coated fruit.
41

42 A12: Many thanks for your careful observation. We have added the information in the text as
43 presented below: *Aloe arborescens* leaves were harvested from plants about 6-year old.
44
45

46 Q13: The gel was applied alone, directly to the fruit?. They did not consider any elements?. It is
47 necessary to argue strongly this point. There is abundant scientific evidence regarding the use of
48 Aloe with plasticizers and emulsifiers in a formulation applied by immersion and subsequent
49 drying.
50

51 A13: Many thanks for your careful observation. Many works in the literature on the use of edible
52 coatings suggest the use of additives such as antioxidants... to improve the product's shelf-life. In
53 our experiment we decided to use 100% *aloe arborescens* to make an edible gel, since almost all
54 the literature uses *aloe vera* (also on strawberries). Thus, in order to evaluate fully the effects of
55 the *aloe arborescens* gel, we chose not to use additives.
56
57

58 Q14: It is necessary to incorporate information regarding procedures on firmness determination,
59 one or two faces of the fruit?
60

A14: We thank fort this comment. Necessary information was added to materials and methods

1
2
3 section as presented below:

4 Fruit firmness was measured as total firmness by a compression test. Each sample was subjected
5 to a two-cycle compression with 5 s between cycles. The highest value of force required to
6 compress the sample during the first compression cycle was recorded as fruit firmness of
7 strawberry. This test measured individual fruit firmness based on the resistance of the flesh to
8 deformation by the probe. Firmness was measured at the equatorial part and on two faces of
9 each strawberry fruit.
10
11

12
13 Q15: It is necessary to incorporate information regarding the experimental strategy and statistical
14 design considered, design completely random, factorial, etc., these aspects are not mentioned in
15 the manuscript that only shows partially this information related to the treatment of the results.
16 The research does not have statistical design and conclusion in this aspect.

17 A15: Many thanks for your careful observation.

18 The main aim of the present work was that of evaluating the effect of the *aloe arborescens* coating
19 on the shelf life of freshly-picked strawberries which were immediately taken to the laboratory. For
20 this reason we did not take into account the experimental strategy or statistical design. We based our
21 choice on a selection of commercially ripe fruit which showed a uniform red surface.
22 As written in Materials and Methods, the strawberries were randomly hand-picked.

23
24
25

26 Q16: It would be very valuable that the article deepens regarding the possible mechanisms of the
27 alteration of the sugar content in function of hydrolysis of polysaccharides in function of the
28 treatment applied.
29

30 A16: Many thanks for your careful observation. Unfortunately, we did not consider this aspect of
31 particular importance. We will certainly bear this suggestion in mind in future studies.
32

33 Q17: Line 277: falta referencia bibliografica que respalde la afirmación.

34 A17: Line 277 now line 286 reported our results. No reference is necessary.
35
36

37 Q18: Line 284: "Intakes for Vitamin C, Vitamin E, Selenium, and Carotenoids, 2000)". Falta signo
38 coma antes del año de la referencia.
39

40 A18: Comma was inserted before year.
41

42 Q19: Line 293. Señalan perdida de componentes en tratamiento control, por alza en respiration
43 pero ellos no lo midieron. Esto debilita el peso de las posibles conclusiones a desarrollar.
44

45 A19: Thank you for comments. Taking into account that respiration rate was not evaluated the
46 comment reported in line 293 was removed as well as Reference Ali et al. (2016).
47

48 Q20: Line 201: "color. The use of edible coatings such as *A. arborescens* gel make a layer on
49 strawberries skin.... Missing letter in scientific number.
50

51 A20: We have checked and corrected.
52

53 Q21: In References section: Page15, line 53, the references must have one unique format (see
54 Line 361 and 365).
55

56 A21: We have checked and corrected journal abbreviation as follow: "*LWT - Food Sci. Technol.*"
57

58 Q22: In section Figures: Figures1 is very small. It does not illustrate the process of extraction or
59 coating on the fruit. It could be dispensed with in the current state, since it does not provide
60 relevant information for the manuscript.

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2
3 **A22: Following the Referee suggesting we have removed Figure 1.**
4

5 **Referee: 3**
6

7
8 Q1: Differences in selected properties of 3 groups of strawberries were determined on 6 different
9 days during a period of 14 days storage at 3 °C and 90%? relative humidity, in order to assess the
10 efficiency of a coating with a gel of *Aloe arborescens*. However, discussion did not consider the
11 actual variation of the selected properties, which are only shown in 2 tables and 1 figure.
12

13 **A1: We have improved discussion section.**
14

15 Q2. English language editing is necessary. There are also a lot of print errors.

16 **A2: Native English speaker checked manuscript for editing. Moreover manuscript was checked
17 for typos.**
18

19
20 Q3: 19: "... stored in refrigerator at 3 °C and 50 % relative humidity..." and in lines 110-111: "The
21 three different batches of containers (C, PP, EC) were all refrigerated at 3 ± 1 °C (90 % RH)."
22 Consider revising.
23

24 **A3: The required corrections were carried out.**
25

26 Q4: lines 94-95: "The gel obtained from the leaves was filtered to remove fibres and pasteurised
27 at 70 °C for 45 minutes." What kind of filter was used? Was it filtration or sieving?
28

29 **A4: We thank the referee for this comment. Necessary information was added to materials and
30 methods section as presented below:**

31 **The obtained matrix was filtered, to remove fibrous fraction, under vacuum filtration with Buchner
32 funnel using laboratory filter paper. The gel was pasteurized in a thermostatic bath at 70 °C (core)
33 for 45 minutes (Arowora *et al.*, 2013), and allowed to cool immediately to an ambient temperature
34 before further use. The treatment was monitored using a Data Logger (Escort Junior, Astori
35 tecnica, Italy).**
36
37

38 Q5: How was pasteurisation performed, in a water bath? Please be precise.
39 109-110: "... immersing fruits for 1 min in *Aloe arborescens* gel. After dipping fruits were allowed
40 to air dry for 60 minutes using a high-speed fan." How was the fruit displayed for drying at what
41 temperature? Was there any surface contact of the fruits with other material (wood, plastic)?
42

43 **A5: We thank fort this comment. Necessary information was added to materials and methods
44 section as presented below: The treatment with EC was done by immersing fruits for 1 min at room
45 temperature in *Aloe arborescens* gel. Based on our previous studies, longer dipping times showed
46 no change in results (data not shown). After dipping the fruits were places in stainless steel trays
47 and to airdried for 60 minutes at room temperature for 15 °C. All forty-two samples, subdivide as
48 described, were refrigerated at 3 ± 1 °C (90 % RH).**
49
50

51
52
53 Q6: line 137: "Determination of pH, total soluble solid and ascorbic acid content". What about
54 titrable acidity? Was it the pH of crushed fruits or a dilution?

55 **A6: Many thanks for your careful observation. The missed part regards to titrable acidity was
56 added properly to the materials and methods section as showed below:**

57 **Determination of pH, titratable acidity, total soluble solid and ascorbic acid content**

58 **Strawberries (c.a. 5 g) were ground in a commercial blender, added with 50 mL of H₂O and
59 homogenized using Ultraturrax T-25 (Ika Labortechnik, Janche & Kunkel, Milan, Italy). The mix was**
60

1
2
3 centrifuged at 5000 rpm for 10 min and the supernatant was collected and use for all
4 determination.
5

6
7 Q7: lines168-169: "... TAC content ..." Redundancy.

8 A7: The method was reported and the redundancy was eliminated.
9

10
11 Q8: 318-319: "Pearson's correlation coefficient evidenced that TPC and Ascorbic acid values are
12 positively linked to the founded antioxidant activity." No correlation results have been reported in
13 manuscript.

14 A8: Following Referee suggestion we have decided to remove Pearson's correlation coefficient
15 calculation.
16
17
18

19 We hope the revised manuscript will better suit *the International Journal of Food Science and*
20 *Technology* but are happy to consider further revisions, and we thank you for your continued
21 interest in our research.
22

23
24 Sincerely,

25 Monica Rosa Loizzo
26

27
28 
29

30
31 Associate Professor in Food Science Technology
32 Department of Pharmacy, Health Science and Nutrition
33 University of Calabria
34 Arcavacata Rende (CS), 87036, Italy
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